

8 — Excitation & Ionization [*Revision* : 1.3]

- Excitation

- At absolute zero, all atoms of a given element are stationary, and electrons in each atom are in ground energy level. As temperature raised above zero, atoms undergo **thermal motions** ($mv^2/2 \approx 3kT/2$)
- **Collisions** between moving atoms impart energy to electrons, **exciting** them to higher energy level
- Convention: label energy levels using index j :
 - * $j = 1, 2, 3, \dots$
 - * $j = 1$ is ground level
 - * E_j is energy of j 'th level (j ordered so that $E_{j+1} > E_j$)
 - * N_j is **level population** — number of atoms in level j
- Each energy level can be composed of a number of **degenerate** (equal-energy) quantum states; this number is **statistical weight** g_j of level
- Boltzmann statistics: at temperature T , probability an atom in energy level j :

$$P_j \propto g_j e^{-E_j/kT}$$

- For any pair of levels j and j' , ratio of probabilities:

$$\frac{P_{j'}}{P_j} = \frac{g_{j'}}{g_j} e^{-(E_{j'}-E_j)/kT}$$

- Equivalent to ratio of level populations:

$$\frac{N_{j'}}{N_j} = \frac{P_{j'}}{P_j} = \frac{g_{j'}}{g_j} e^{-(E_{j'}-E_j)/kT}$$

- Because $\sum_{j=1} P_j = 1$,

$$P_j = \frac{g_j e^{-(E_j-E_1)/kT}}{Z}$$

where Z is **partition function**:

$$Z = \sum_{j=1} g_j e^{-(E_j-E_1)/kT}$$

- Ionization

- If enough energy given by collision, electron can be removed from atom — **ionization**
- Convention: label ionization stages using index i :
 - * $i = 1, 2, 3, \dots, Z - 1$ (where Z is atomic number)
 - * $j = 1$ is neutral (unionized) stage
 - * $E_{i,j}$ is energy of j 'th level in i 'th ionization stage
 - * $g_{i,j}$ is corresponding statistical weight
 - * χ_i is **ionization potential** of i 'th stage — amount of energy to remove a further electron, and produce $i + 1$ 'th stage
 - * $N_{i,j}$ is number of atoms in stage i and level j
 - * $N_i = \sum_j N_{i,j}$ is total number of atoms/ions in stage i

- Consider ionization but ignore excitation: all atoms/ions in ground level ($j = 1$)
- Ionization process: removing single electron from atom/ion in stage i , to produce ion in stage $i + 1$ plus free electron. Can apply Boltzmann statistics — trick is to calculate statistical weights correctly
- For process producing free electron with momentum in interval $(p_e, p_e + dp_e)$, ratio between numbers:

$$\frac{dN_{i+1}}{N_i} = \frac{g_{i+1,1} dg_e}{g_{i,1}} e^{-(\chi_i + p_e^2/2m_e)/kT}$$

(Note: energy term is sum of ionization potential χ_i and electron kinetic energy $p_e^2/2m_e = m_e v_e^2/2$). Electron statistical weight dg_e is number of quantum states available to electron with momentum in interval $(p_e, p_e + dp_e)$

- To calculate dg_e , use same approach as with blackbody radiation (see Notes 3):
 - * Wave-particle duality: treat electrons as waves, number of states \leftrightarrow number of permitted waves
 - * In box with dimensions dV , number of standing waves in wavenumber interval $(k, k + dk)$:

$$dg_e = \frac{k^2 dk}{\pi^2} dV$$

(Note: same expression as BB notes, but different notation. Also, in derivation hidden factor of 2 for light polarizations is replaced by factor of 2 for electron spins)

- * Use de Broglie relation $p_e = \hbar k = \hbar k/2\pi$:

$$dg_e = \frac{8\pi p_e^2 dp_e}{h^3} dV$$

- * To find dV : if there are n_e electrons per unit volume, then each electron has $dV = 1/n_e$ available to it:

$$dg_e = \frac{8\pi p_e^2 dp_e}{h^3 n_e}$$

(Note: this is a classical-physics fudge, the proper way is to use **Fermi-Dirac statistics** from the start)

- So:

$$\frac{dN_{i+1}}{N_i} = \frac{g_{i+1,1}}{g_{i,1}} \frac{8\pi p_e^2 dp_e}{h^3 n_e} e^{-(\chi_i + p_e^2/2m_e)/kT}$$

- Integrate over all electron momenta to get total number in stage $i + 1$:

$$\frac{N_{i+1}}{N_i} = \int \frac{dN_{i+1}}{N_i} = \frac{g_{i+1,1}}{g_{i,1}} \frac{8\pi}{h^3 n_e} e^{-\chi_i/kT} \int_0^\infty p_e^2 e^{p_e^2/2m_e kT} dp_e$$

- Use identity

$$\int_0^\infty x^2 e^{-a^2 x^2} dx = \frac{\sqrt{\pi}}{4a^3}$$

to get result

$$\frac{N_{i+1}}{N_i} = \frac{2g_{i+1,1}}{n_e g_{i,1}} \left(\frac{2\pi m_e kT}{h^2} \right)^{3/2} e^{-\chi_i/kT}$$

- Additional correction: allow for excitation amongst energy levels of each ionization stage \rightarrow replace $g_{i,1}$ by Z_i and $g_{i+1,1}$ by Z_{i+1} :

$$\frac{N_{i+1}}{N_i} = \frac{2Z_{i+1}}{n_e Z_i} \left(\frac{2\pi m_e kT}{h^2} \right)^{3/2} e^{-\chi_i/kT}$$

This is the famous **Saha equation** (sometimes called Saha-Boltzmann equation)